

# Optical characterization of atmospheric aerosols on quartz filters using reflectance-based spectroscopy

Michael J. Tackett Jr.

University of Arkansas Little Rock

Department of Chemistry

Department of Earth Science

# Why study aerosol absorption?

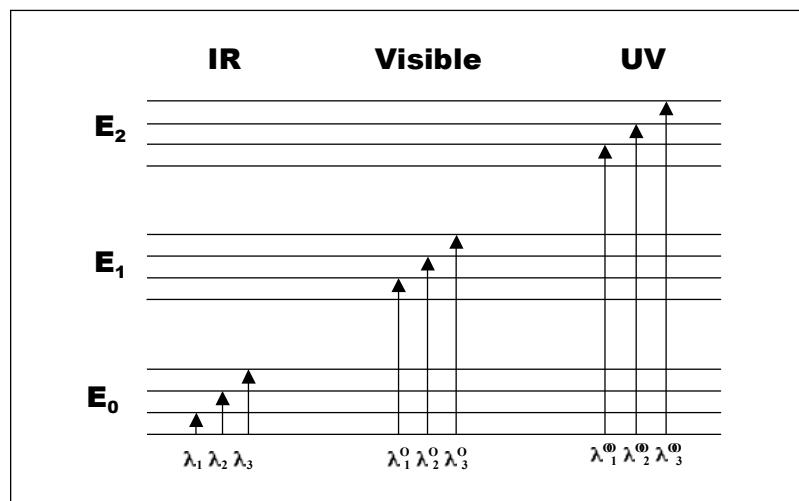
- o Atmospheric aerosols can both scatter **AND** absorb
- o Aerosol scattering mainly correlated with average particle size; **absorption** correlated with **chemical composition**
- o **Black carbon (BC)** aerosols have the ability to scatter light, causing cooling, while simultaneously absorbing solar energy, causing:
  - o Significant local warming of boundary layer
  - o Similar warming in lower atmosphere as major greenhouse gases
- o Important in **UV** for photochemical modeling

# Background: Absorption Spectroscopy

- o **Spectroscopy**=study of any quantity as a function of wavelength ( $\lambda$ ) or frequency ( $\nu$ ); most often, interaction of radiation with matter
- o **Overall energy (E)** of molecule described by:

$$E_{\text{molecule}} = E_{\text{electronic}} + E_{\text{vibrational}} + E_{\text{rotational}} + E_{\text{translational}}$$

- o **UV/Vis=electronic** transitions
  - o promotion of  $e^-$  from ground state to higher energy orbitals



# Background: Absorption Theory

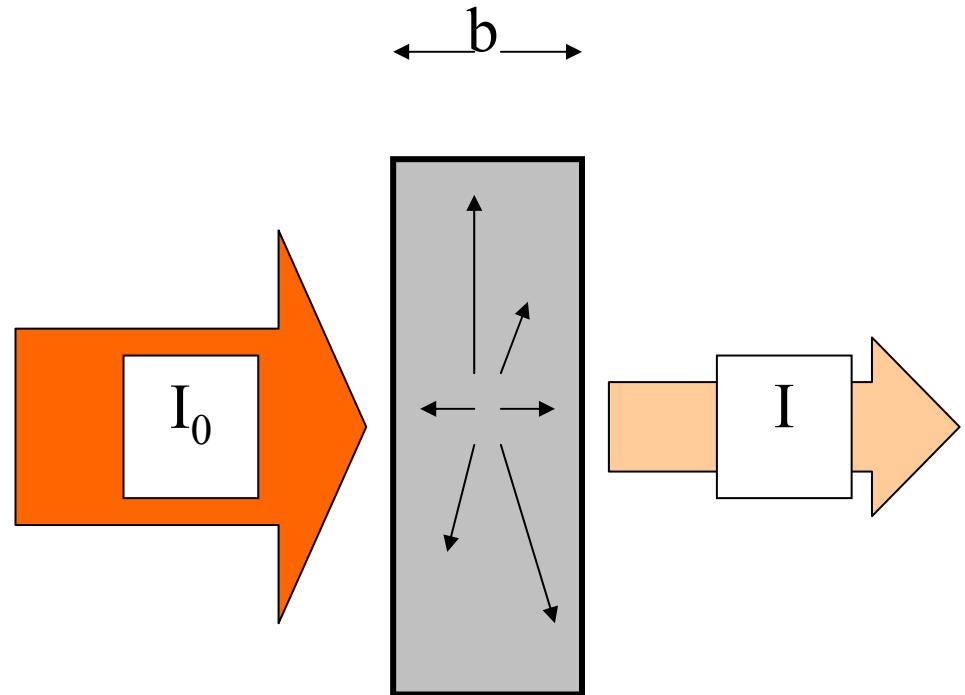
- o **Beer-Lambert Law:**

$$A = \epsilon bc$$

- o **Measured property:  
transmission;**

$$T = (I/I_0);$$

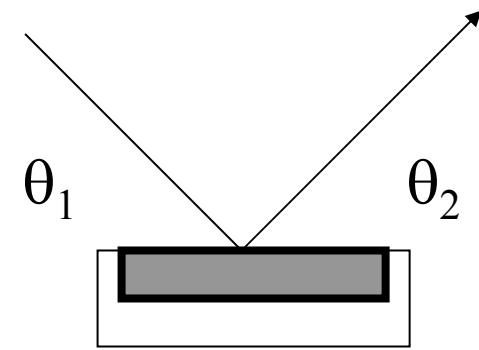
$$A = \log(1/T) \text{ or } -\log(T)$$



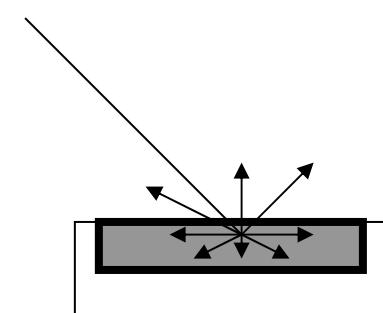
Sample

# Background: Reflection Theory

- o **Used** when transmission through sample not feasible/measurable
- o **2 types of reflections:**
  - Specular**=mirror-type ( $\theta_1 = \theta_2$ )
  - Diffuse**=scattering in all directions
- o **Kubelka-Munk Theory**=developed for paper industry; dyed paper;  
Kubelka-Munk function ( $f(K - M) = \frac{(1 - R)^2}{2R}$ ) linearizes reflectance data by applying scattering factor; corrects for extra path length from scattering within sample



Specular

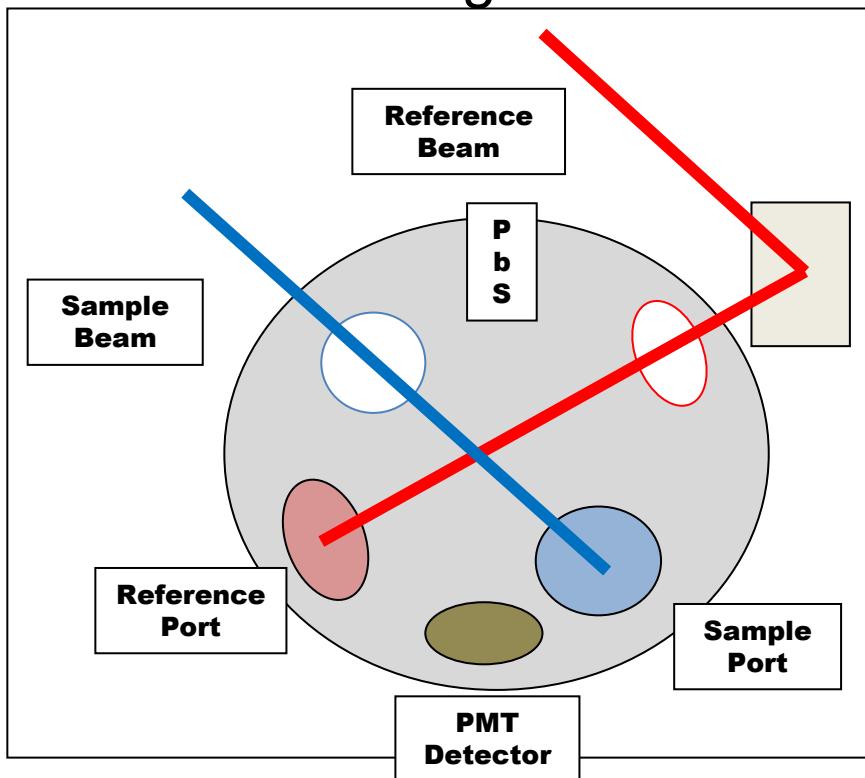


Diffuse

# Instrumentation

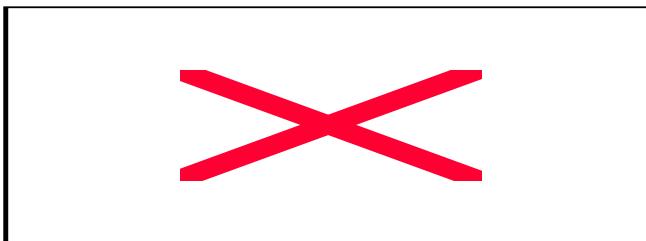
- o Perkin-Elmer Lambda 850 UV/Vis Spectrometer with 150mm Labsphere Integrating Sphere
  - o Integrating Sphere-corrects for scattering off surface of sample

\*ability to measure transmittance/reflectance at all wavelengths across the spectrum (250-850 nm)



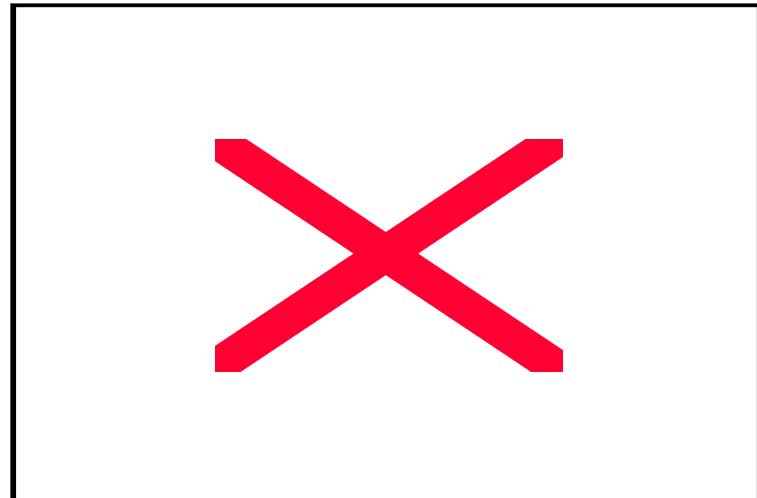
# Instrumentation (cont'd)

- o Particle Soot/Absorption Photometer (PSAP)



\*filters then taken for  
spectroscopic analysis

\*measures only at 1 to 3 wavelengths



- o High-volume aerosol impactors

# Absorption Angström Exponents (AAEs)

- o **Absorption Angström Exponent (AAE;  $\alpha$ ):** describes wavelength dependence of aerosol absorbance; controlled mainly by chemical composition; results from combination of broadband (BC, HULIS) and narrowband (PAHs, N-PAH) absorption

- o 
$$A = \beta \lambda^{-\alpha}$$

where A=measured absorbance

$\beta$ =Absorption at 1  $\mu\text{m}$  (1000 nm)

$\lambda$ =Wavelength (nm)

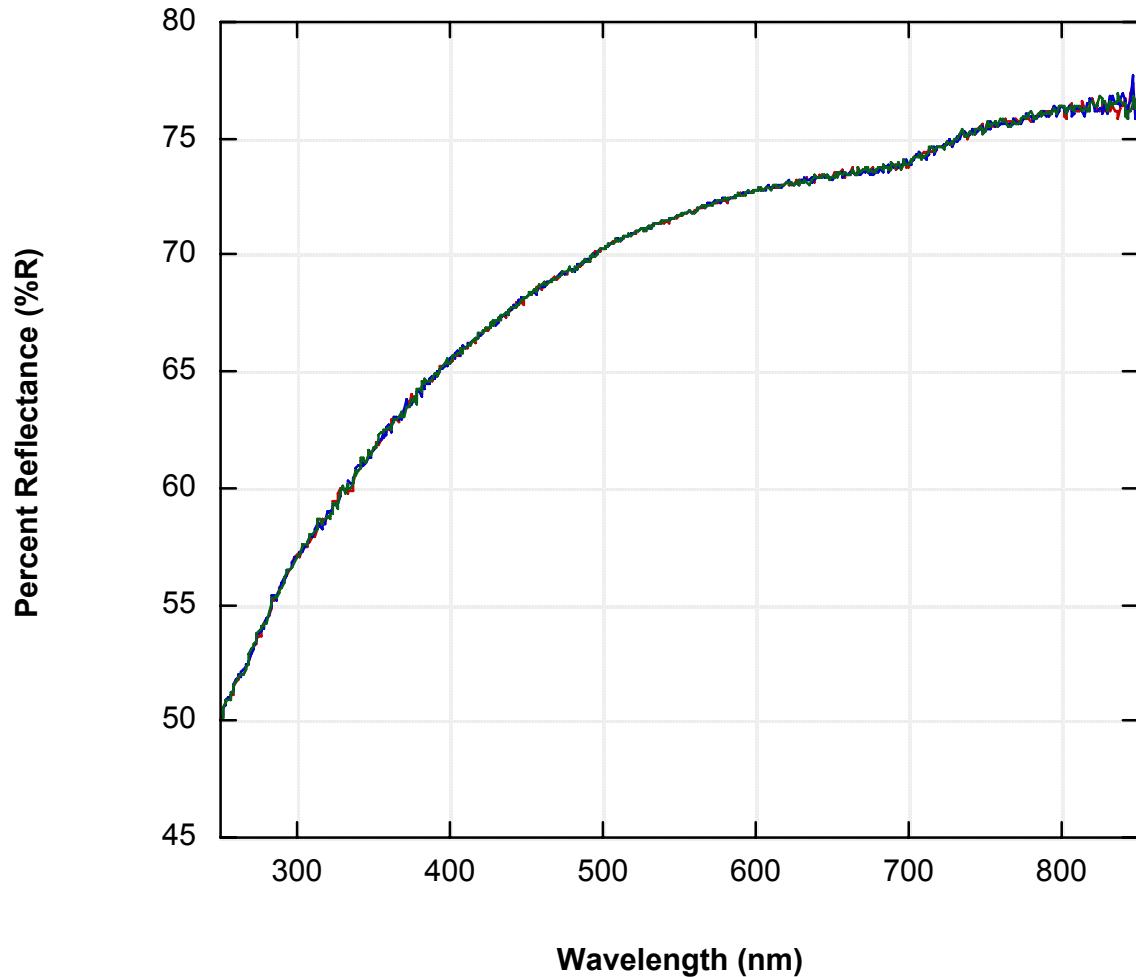
$\alpha$ =Absorption Angström Exponent (AAE)

- o **To solve for the AAE,** take the natural log of both wavelength and absorbance values; plot graph of Ln of absorbance as a function of Ln of wavelength; run linear regression; AAE = slope (m) of the linear equation

# UV/Vis Data (Mount Bachelor)

Percent Reflectance Cycle 1  
Percent Reflectance Cycle 2  
Percent Reflectance Cycle 3

Percent reflectance versus  
wavelength,  
PSAP 041808

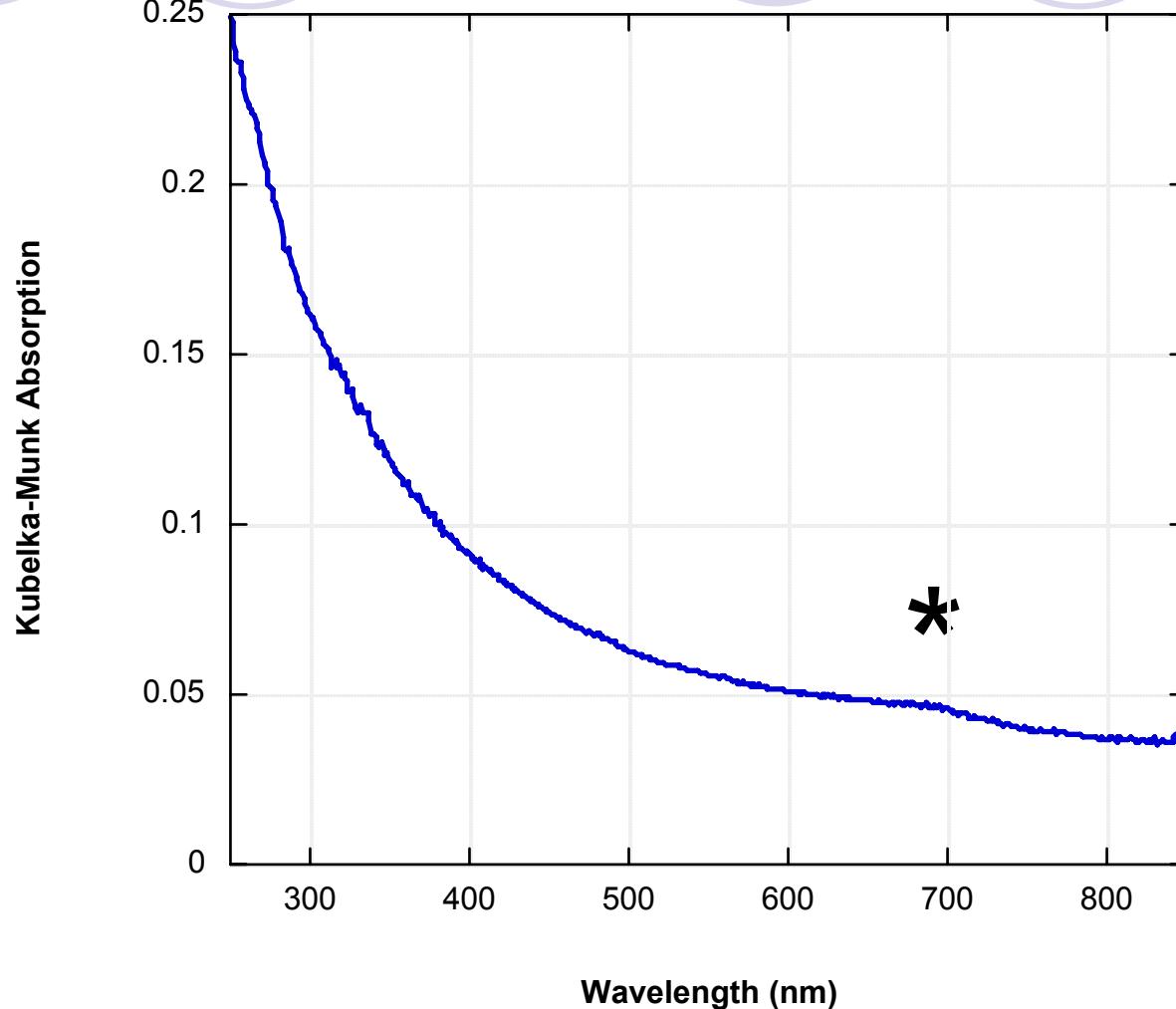


- Complete example of calculating AAE for April 18, 2008

\*very small standard deviation

— Kubelka-Munk Absorption

### Kubelka-Munk Absorbance versus wavelength, PSAP 041808

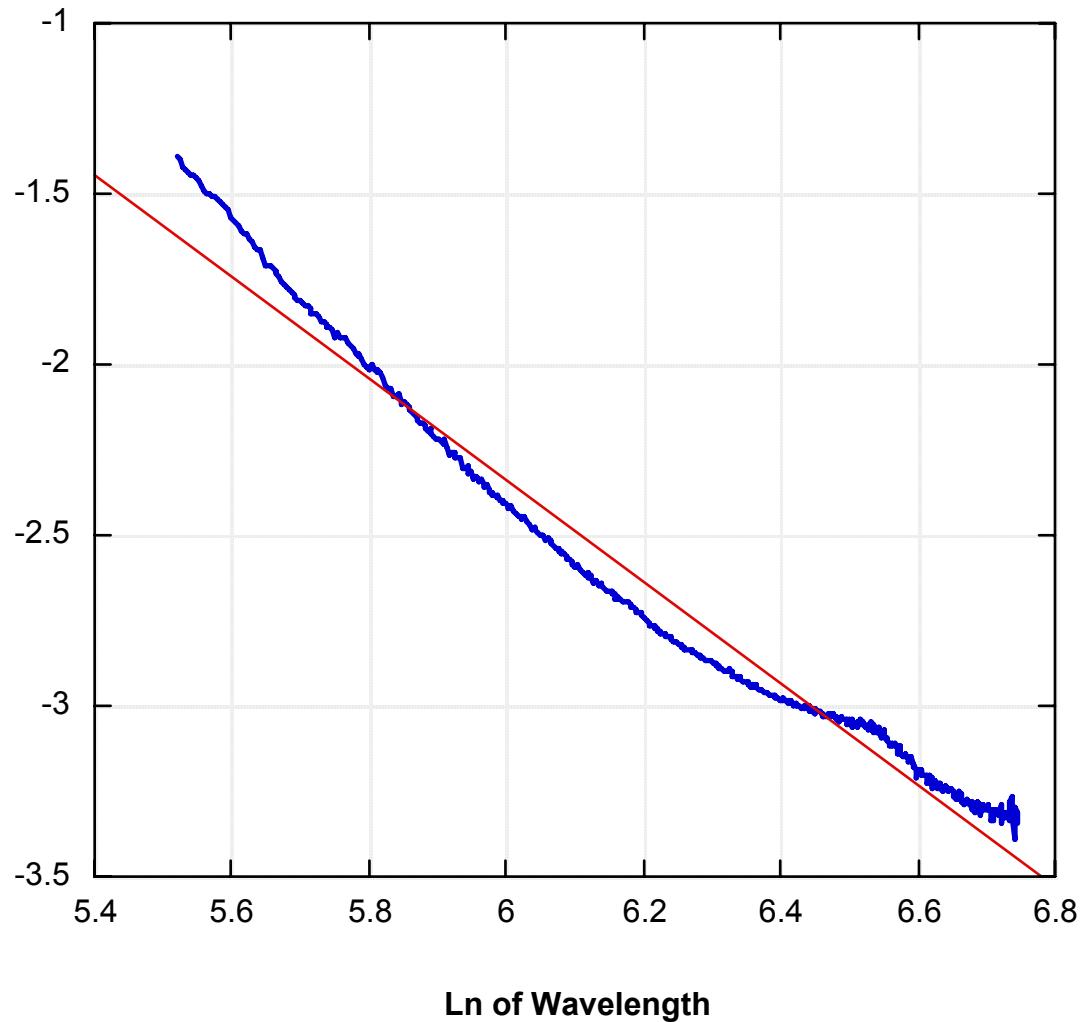


\* Example of narrowband absorption within broadband absorption

— Ln of Kubelka-Munk Absorption

Natural log of Kubelka-Munk Absorption versus  
natural log of wavelength,  
PSAP 041808

Ln of Kubelka-Munk Absorption



$$y = 6.6 - 1.5x$$

$$R^2=0.974$$

$$\alpha=1.5$$

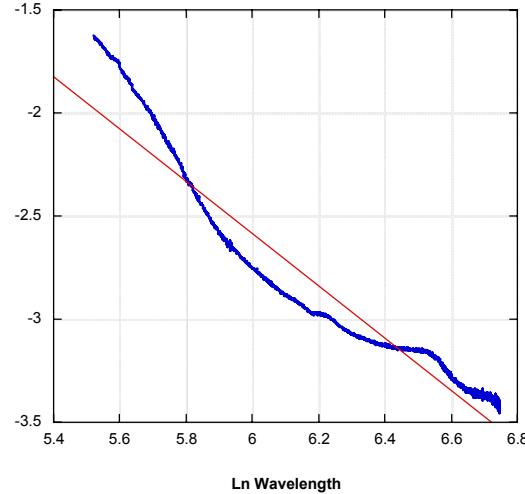
# UV/Vis Data (Mount Bachelor) cont'd

## o PSAP Spectra (Mt. Bachelor; March-May, 2008)

—

Ln of Kubelka-Munk Absorption

Natural log of Kubelka-Munk Absorption versus  
natural log of wavelength,  
PSAP 032908



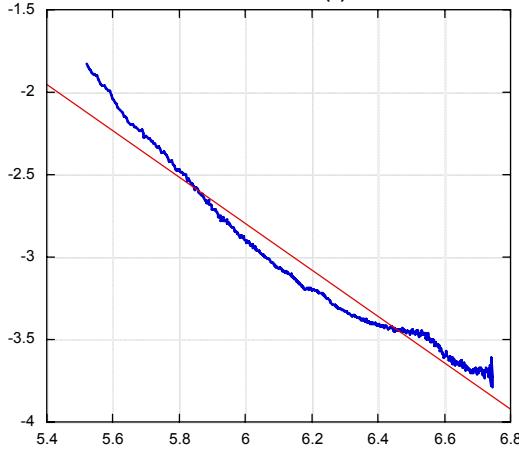
$$R^2=0.915$$

$$\alpha=1.3$$

—

Ln of Kubelka-Munk Absorption

Natural log of Kubelka-Munk Absorption versus  
natural log of wavelength,  
PSAP 040908(1)



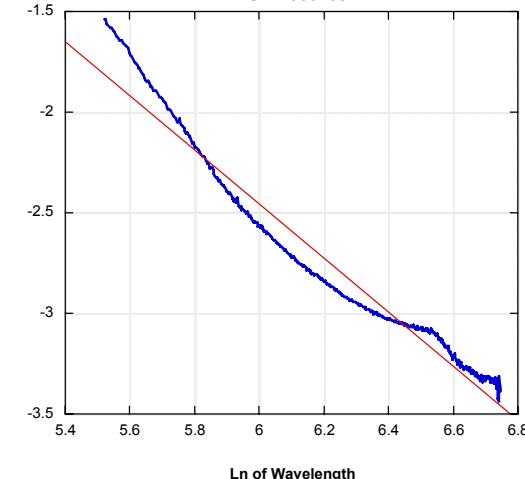
$$R^2=0.958$$

$$\alpha=1.4$$

—

—

Natural log of Kubelka-Munk Absorption versus  
natural log of wavelength,  
PSAP 050108



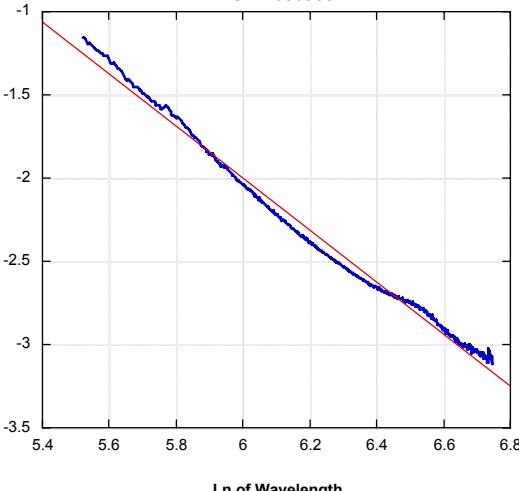
$$R^2=0.958$$

$$\alpha=1.3$$

—

—

Natural log of Kubelka-Munk Absorption versus  
natural log of wavelength,  
PSAP 050908

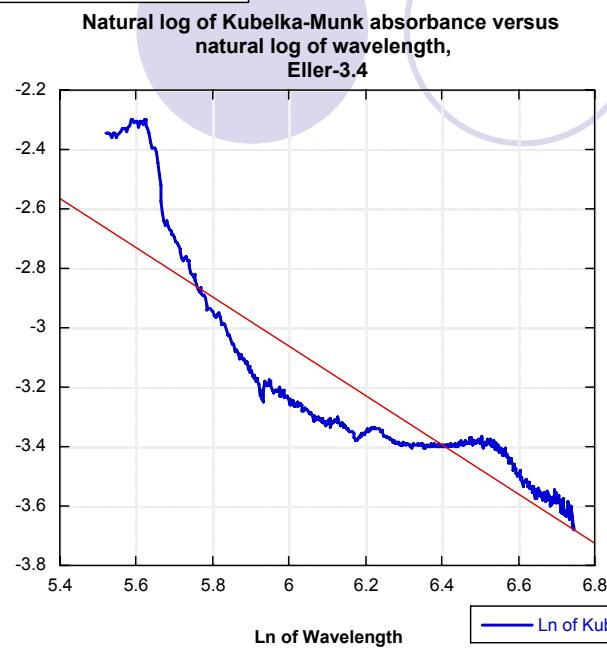


$$R^2=0.991$$

$$\alpha=1.6$$

## o High-Volume Impactor Spectra (UALR)

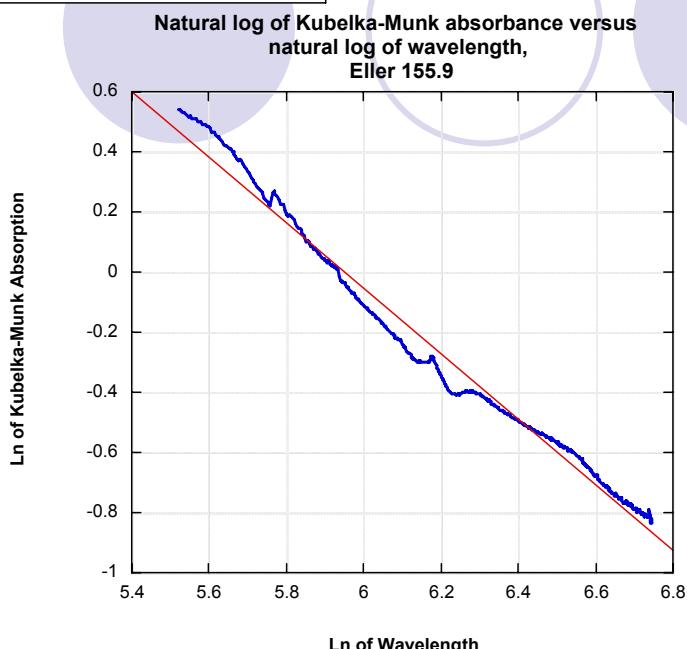
— Ln of Kubelka-Munk Absorption



$$R^2 = 0.801$$

$$\alpha = 0.83$$

— Ln of Kubelka-Munk Absorption



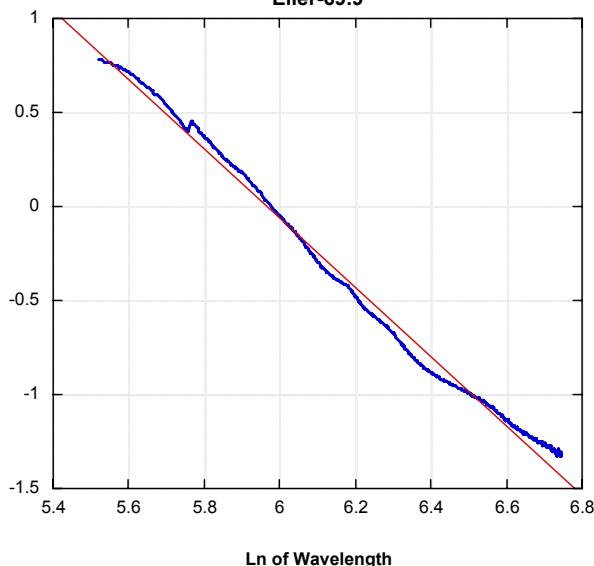
$$R^2 = 0.982$$

$$\alpha = 1.1$$

— Ln of Kubelka-Munk Absorption

Natural log of Kubelka-Munk absorbance versus  
natural log of wavelength,  
Eller-69.9

Ln of Kubelka-Munk Absorbance

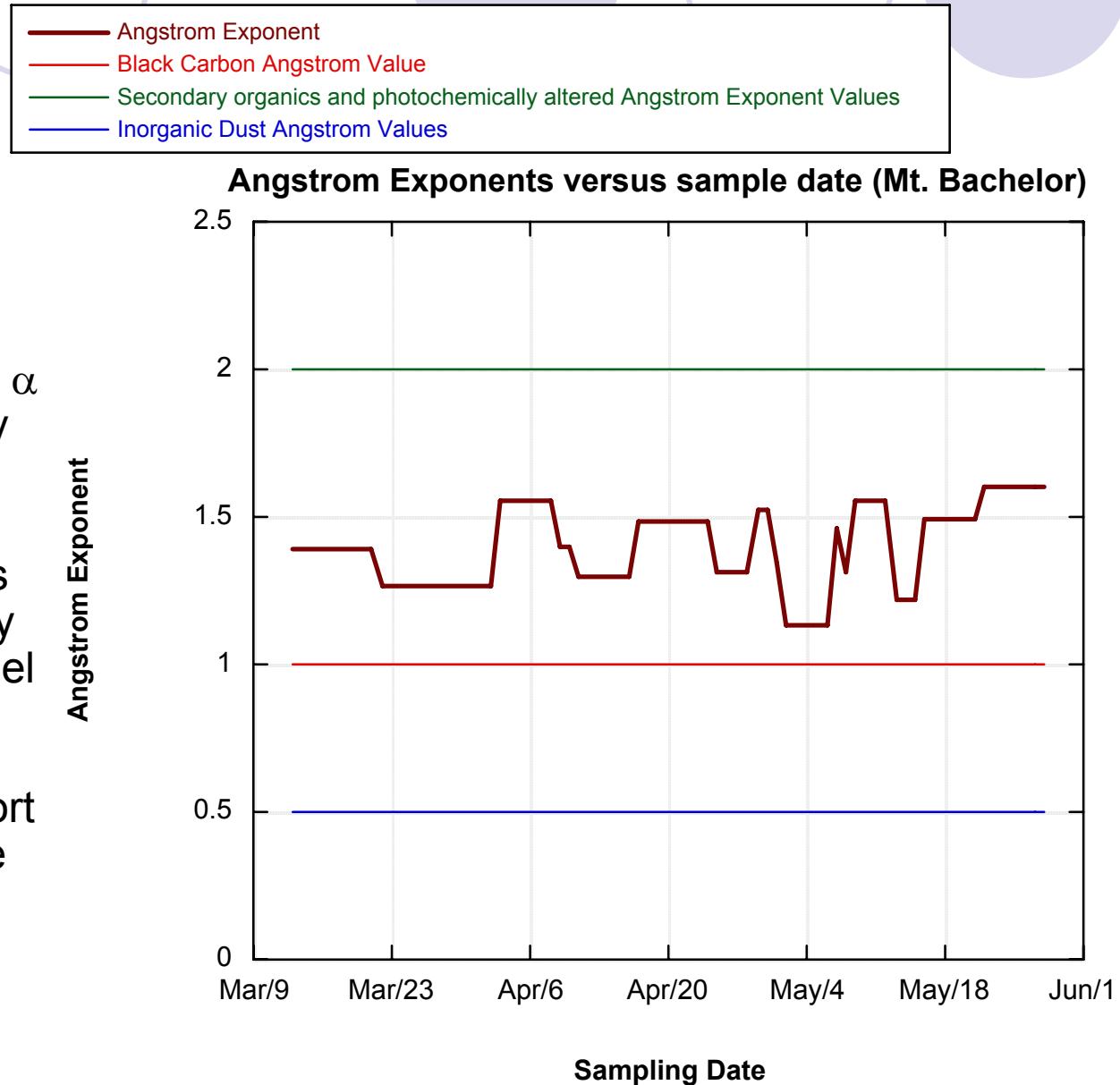


$$R^2 = 0.992$$

$$\alpha = 1.8$$

# UV/Vis: Conclusions (Mt. Bachelor)

- o Series of short influxes of maybe slightly photochemically-altered BC aerosols, causing  $\alpha$  to become slightly larger
- o Long-range transport aerosols from Asia=primary organics (i.e. diesel soots) with little photochemical alteration; transport over ocean where other organic reactants not present



# Future Work

- o Begin to correlate with aethalometer absorption data
- o Run samples using Fourier-Transform Infrared Spectroscopy (FTIR)
- o Continue to run UALR impactor samples
- o Hopefully soon!: Assemble an in-line combustion-GC system for EC/OC analysis of filters post-spectroscopy

# References

Skoog, West, Holler, and Crouch. *Analytical chemistry: An introduction.* 7th Ed. 2000. Brooks/Cole

Labsphere 150mm Integrating Sphere Owner's Manual

Nancy A. Marley, Jeffrey S. Gaffney, Michael Tackett, Neil C. Sturchio, Linnea Heraty, Nancy Martinez, Kavita D. Hardy, Angie Machany-Rivera, Thomas Guilderson, Amanda MacMillan, and Karen Steelman. "The Impact of Biogenic Carbon Emissions on Aerosol Absorption in Mexico City." ACP, Special MILAGRO issue. Submitted.

# Acknowledgments

- o Dr. Jeff Gaffney (mentor; academic/research advisor)
- o Dr. Nancy Marley (mentor; research advisor)
- o Emily Fischer (GREF student; PSAP samples)
- o Milton Constantin & team
- o GCEP/SURE
- o DOE and ASP
- o Dr. Amy Robinson (academic advisor)

## Extras

- o Angstrom Exponent as function of sampling duration

